

## Lesson Plan 2024(summer)

Subject :-ENGINEERING CHEMISTRY (Code) Th.2b.

Name of faculty:- LIPSHARANI BARIK

Semester :-2<sup>nd</sup>

Class allotted 4p/w

Branch :-MECHANICAL & METALLURGY

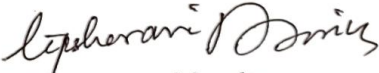
Discipline	Semester:-2 <sup>nd</sup>	From date:-29/01/24 To date:14/05/24	Teaching Aid
Subject:	No. of days/ week 4p/w	Theory/ Practical –Topics/Lesson	
Week	Date		

1	29/01/24 to 03/02/24	<b>A. PHYSICAL CHEMISTRY</b> <b>Chapter 1: Atomic structure</b> : Fundamental particles ( electron, proton & neutron Definition, mass and charge ).Rutherford's Atomic model ( postulates and failure), Atomic mass and mass number, Definition, examples and properties of Isotopes, isobars and isotones. Bohr's Atomic model ( Postulates only), Bohr-Bury scheme, Aufbau's principle, Hund's rule	White board & marker
2	05/02/24 to 10/02/24	Electronic configuration (up to atomic no 30). <b>Chapter 2 : Chemical Bonding</b> : Definition , types ( Electrovalent, Covalent and Coordinate Bond with examples ( formation of NaCl, MgCl <sub>2</sub> , H <sub>2</sub> ,Cl <sub>2</sub> , O <sub>2</sub> , N <sub>2</sub> , H <sub>2</sub> O, CH <sub>4</sub> , NH <sub>3</sub> , NH <sub>4</sub> <sup>+</sup> , SO <sub>2</sub> ).	White board & marker
3	12/02/24 to 17/02/24	<b>Chapter 3 : Acid base theory</b> : Concept of Arrhenius, Lowry Bronsted and Lewis theory for acid and base with examples ( Postulates and limitations only). Neutralization of acid & base. Definition of Salt, Types of salts ( Normal, acidic, basic, double, complex and mixed salts, definitions with 2 examples from each).	White board & marker
4	19/02/24 to 24/02/24	<b>Chapter 4: Solutions</b> : Definitions of atomic weight, molecular weight, Equivalent weight. Determination of equivalent weight of Acid, Base and Salt. Modes of expression of the concentrations ( Molarity , Normality & Molality) with Simple Problems. pH of solution ( definition with simple numericals ) Importance of pH in industry ( sugar, textile, paper industries only)	White board & marker
5	26/02/24 to 02/03/24	<b>Chapter 5 : Electrochemistry</b> : Definition and types ( Strong & weak) of Electrolytes with example. Electrolysis ( Principle & process) with example of NaCl (fused and aqueous solution). Faraday's 1st and 2 <sup>nd</sup> law of Electrolysis ( Statement, mathematical expression and Simple numerical) Industrial application of Electrolysis- Electroplating ( Zinc only). <b>Chapter 6 : Corrosion</b> : Definition of Corrosion	White board & marker

	04/03/24 to 09/03/24	Types of Corrosion- Atmospheric Corrosion, Waterline corrosion. Mechanism of rusting of Iron only. Protection from Corrosion by (i) Alloying and (ii) Galvanization. <b>B. INORGANIC CHEMISTRY</b> <b>Chapter 7 : Metallurgy:</b> Definition of Mineral, ores , gangue with example. Distinction between Ores And Minerals.	White board & marker
7	11/03/24 to 16/03/24	General methods of extraction of metals, i) Ore Dressing ii) Concentration ( Gravity separation, magnetic separation, Froth floatation & leaching) iii) Oxidation (Calcinations, Roasting ) iv) Reduction (Smelting, Definition & examples of flux, slag) v) Refining of the metal ( Electro refining, & Distillation only)	White board & marker
8	18/03/24 to 23/03/24	<b>Chapter 8 : Alloys:</b> Definition of alloy. Types of alloys ( Ferro, Non Ferro & Amalgam) with example. Composition and uses of Brass, Bronze, Alnico, Duralumin <b>C. ORGANIC CHEMISTRY</b> <b>Chapter 9 : Hydrocarbons :</b> Saturated and Unsaturated Hydrocarbons ( Definition with example)	White board & marker
9	27/03/24 to 30/03/24	Aliphatic and Aromatic Hydrocarbons ( Huckle's rule only). Difference between Aliphatic and aromatic hydrocarbons IUPAC system of nomenclature of Alkane, Alkene, Alkyne, alkyl halide and alcohol ( up to 6 carbons ) with bond line notation.	White board & marker
10	02/04/24 to 06/04/24	Uses of some common aromatic compounds ( Benzene, Toluene, BHC, Phenol, Naphthalene, Anthracene and Benzoic acid) in daily life.	White board & marker
11	08/04/24 to 13/04/24	<b>D. INDUSTRIAL CHEMISTRY</b> <b>Chapter 10 : Water Treatment :</b> Sources of water, Soft water, Hard water, hardness, types of Hardness (temporary or carbonate and permanent or non-carbonate), Removal of hardness by lime soda method ( hot lime & cold lime— Principle, process & advantages ) , Advantages of Hot lime over cold lime process.	White board & marker
12	15/04/24 to 20/04/24	Organic Ion exchange method ( principle, process, and regeneration of exhausted resins) <b>Chapter 11 : Lubricants:</b> Definition of lubricant, Types ( solid, liquid and semisolid with examples only ) and specific uses of lubricants ( Graphite, Oils, Grease), Purpose of lubrication	White board & marker & smart board
13	22/04/24	<b>Chapter 12 : Fuel:</b> Definition and classification	White board &

	to 27/04/24	of fuel, Definition of calorific value of fuel, Choice of good fuel. Liquid: Diesel, Petrol, and Kerosene --- Composition and uses. Gaseous: Producer gas and Water gas (Composition and uses). Elementary idea about LPG, CNG and coal gas (Composition and uses only).	marker
14	29/04/24 to 04/05/24	<b>Chapter 13 : Polymer:</b> Definition of Monomer, Polymer, Homo-polymer, Co-polymer and Degree of polymerization. Difference between Thermosetting and Thermoplastic, Composition and uses of Polythene, & Poly-Vinyl Chloride and Bakelite.	White board & marker
15	06/05/24 to 11/05/24	Definition of Elastomer ( Rubber). Natural Rubber (it's draw backs ). Vulcanisation of Rubber. Advantages of Vulcanised rubber over raw rubber. <b>Chapter 14: Chemicals in Agriculture:</b> Pesticides: Insecticides, herbicides, fungicides- Examples and uses. Bio Fertilizers: Definition, examples and uses.	White board & marker & smart board
16	13/05/24 to 14/05/24	1)Nomenclature question practice 2)Question Practice 3)Organic chemistry question practice 4)Organic chemistry question practice	White board & marker

  
Signature of HOD

  
Signature of faculty